

CERTIFICATE OF ANALYSIS

Product Name	Urea Analytical standard		
Synonyms	Carbamide, Carbonyldiamide, Ureophil		
CAS No	57-13-6		
Product code	SYI0041	Batch No	LURBF0106
Molecular Formula	CH ₄ N ₂ O	Molecular Weight	60.06
Mfg. Date	Jan-26	Expiry Date	Dec-28
Storage Conditions	Store at room temperature, tightly closed container		

Test	Specification	Results
Description	White powder to crystal	White powder to crystal
Melting Point	132 - 135 °C	135 °C
Solubility	Soluble in Methanol, Ethanol	Complies
Assay	≥99.00%	99.19%
IR Spectrum	Identification by structure	Confirmed
¹H NMR Spectra	Identification by structure	Confirmed
Mass Spectra	Identification by molecular weight	Confirmed

Reference to *USP 30-NF 25* General Chapter <11>, "Reference Standards," As a result, noncompendial (secondary) reference standards require characterization data. This product should not use for clinical application.

Approved by
Dr. Gopinath PH. D

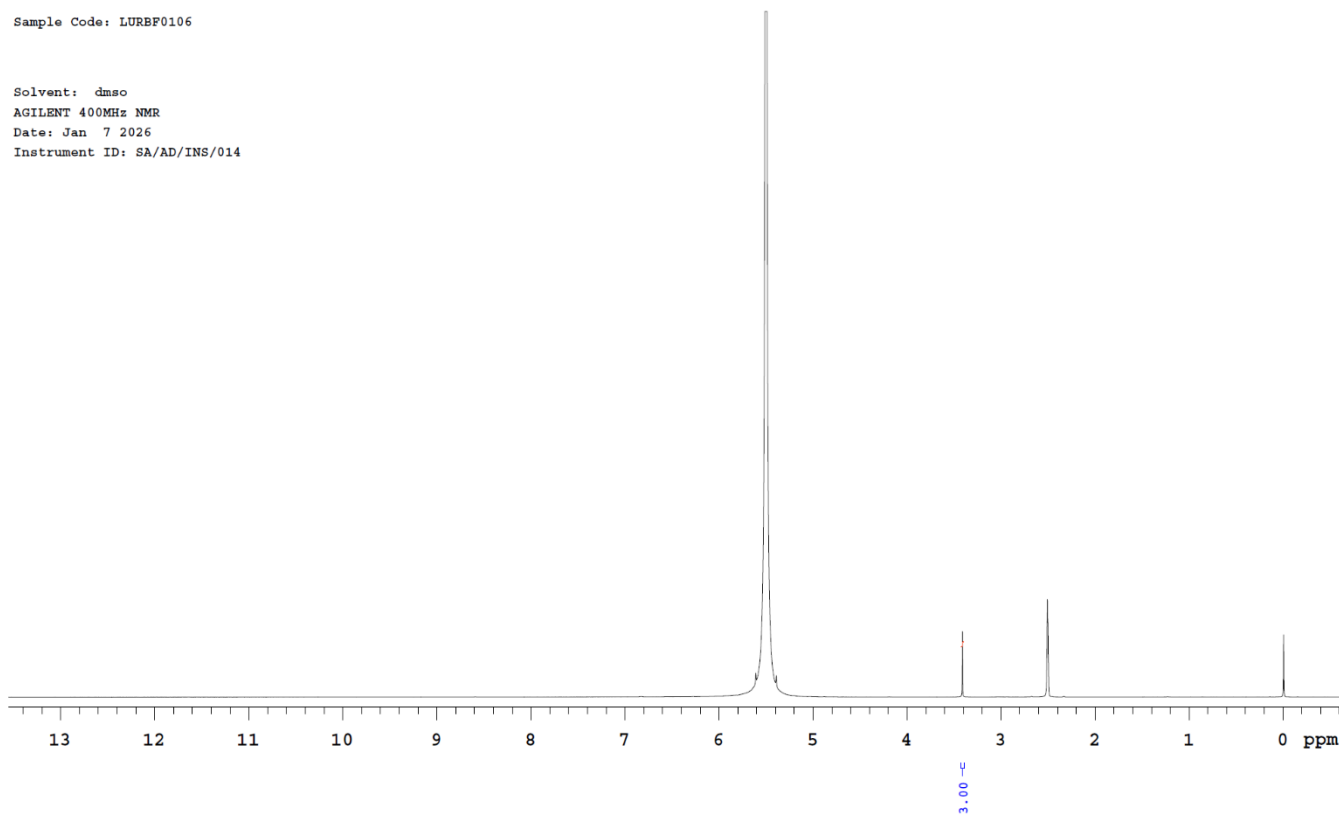


Computer Generated document, Does not require any Signature.

Identification by NMR: H1NMR

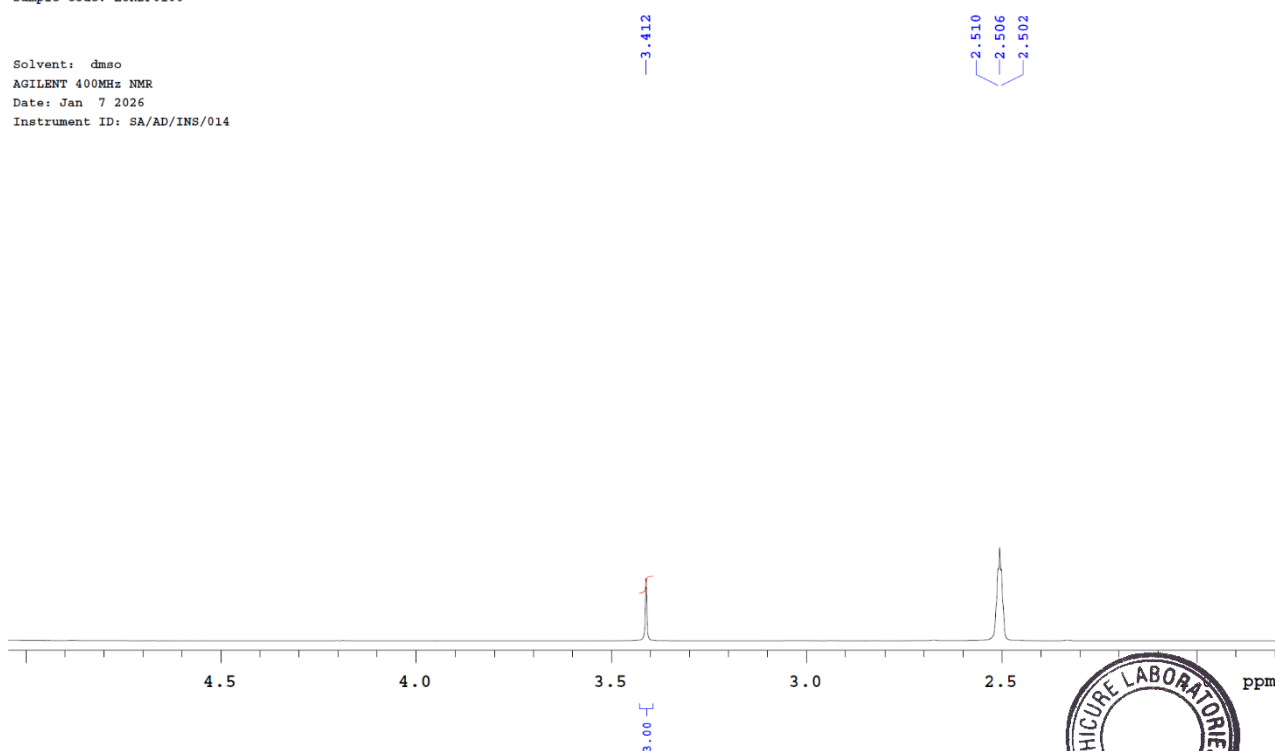
Sample Code: LURBF0106

Solvent: dmsc
AGILENT 400MHz NMR
Date: Jan 7 2026
Instrument ID: SA/AD/INS/014

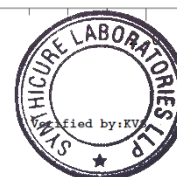


Sample Code: LURBF0106

Solvent: dmsc
AGILENT 400MHz NMR
Date: Jan 7 2026
Instrument ID: SA/AD/INS/014



Plotname: LURBF0106_PROTON_20260107_01_plot02



Identification by NMR: H1NMR

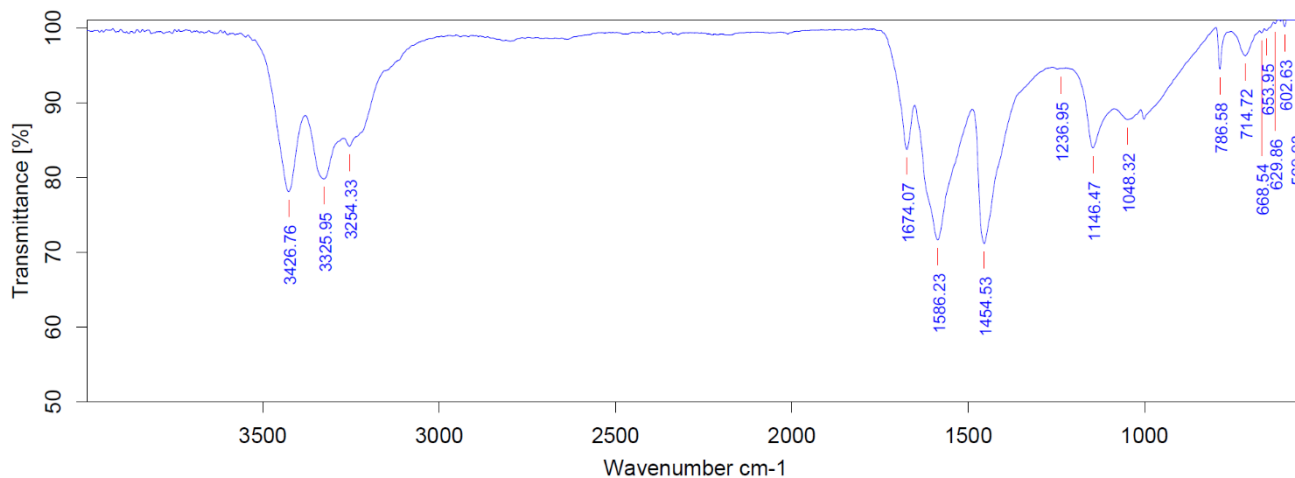
exp2	PROTON	INDEX	FREQUENCY	PPM	HEIGHT
		1	1364.1	3.412	13.1
		2	1003.4	2.510	15.1
		3	1001.9	2.506	19.6
		4	1000.4	2.502	14.9

SAMPLE	PRESATURATION
date Jan 7 2026	satmode n
solvent dms0	wet n
file /home/spark-d-	SPECIAL
d2/data/2026/Jan/L-	temp 25.0
URBF0106_20260107_~	gain 34
01/LURBF0106_PROTO-	spin 20
N_20260107_01.fid	hst 0.008
ACQUISITION	pw90 12.600
sw 7183.9	alfa 10.000
at 4.000	FLAGS
np 57472	il n
fb 4000	in n
bs 2	dp y
d1 1.000	hs nn
nt 128	PROCESSING
ct 16	lb 0.50
TRANSMITTER	fn not used
tn H1	DISPLAY
sfrq 399.829	sp 800.6
tof 799.7	wp 901.3
tpwr 59	rfl 779.4
pw 6.300	rfp 0
DECOUPLER	rp -132.1
dn C13	lp 0
dof 0	PLOT
dm nnn	wc 268
decwave W40_ATB-78-	sc 0
98	vs 147
dpwr 38	th 7
dmf 29412	ai cdc ph

Plotname: LURBF0106_PROTON_20260107_01_plot03



Identification by Infrared Spectroscopy (IR)



Path/File Name: D:\2026\JAN-2026\LURBF0106.0

Sample Name: LURBF0106

Experiment: JANUARY-2026.XPM

Lot No./Batch No: LURBF0106

Resolution: 2

Date & Time: 1/7/2026, 6:17:08 PM

Sample Scans: 32


Operator Name: SPARK

Frequency Range: 4000 to 550

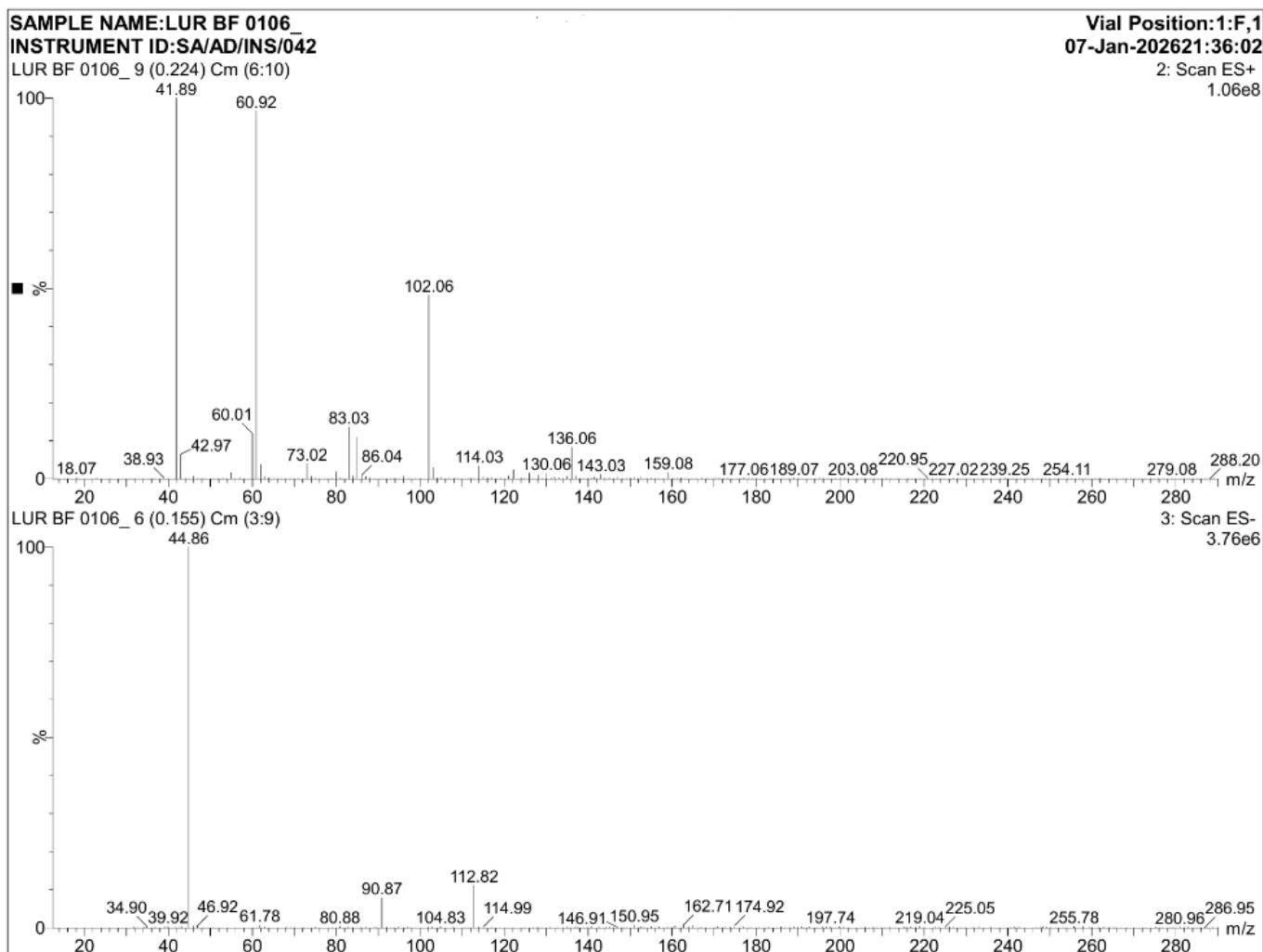
- "D:\2026\JAN-2026\LURBF0106.0" 1
- Peak Table TR
 - Peak Picking

Peak Picking	Values
Method:	Standard
Searched for minima:	Yes
Number of peaks:	16
Sensitivity > [%]:	10.000000
From:	4000.000000
to:	400.000000
Absolute peak height >	0.000000
Relative peak height < [%]	0.000000
Absolute peak height <	0.000000

Wavenumber	Abs. intensity	Rel. intensity	Width	Found if threshold <	Shoulder
3426.7585	0.781	0.218	268.0223	62.497623	0
3325.9545	0.798	0.089	59.1155	24.299952	0
1674.0666	0.837	0.075	25.4299	17.007418	0
1586.2269	0.716	0.179	79.3546	50.026196	0
1454.5301	0.712	0.333	95.2535	82.374474	0
1146.4687	0.840	0.127	143.5856	30.936426	0
786.5807	0.945	0.059	10.8176	15.918880	0
3254.3293	0.842	0.012	16.4704	11.282663	0
1048.3239	0.877	0.027	166.6406	82.034981	0
1236.9467	0.945	0.000	18.9672	0.417842	0
714.7174	0.963	0.047	44.5553	72.073006	0
668.5437	0.994	0.006	2.7208	3.640999	0
653.9480	0.996	0.008	1.8503	2.489824	0
602.6252	1.002	0.033	14.3675	12.617673	0
569.9288	1.008	0.028	19.6139	16.729086	0
529.8569	1.006	0.004	21.8685	2.306005	0



Identification by Mass spectrometry (MS)



Urea purity analysis by Assay Method

Principle: Urea is dissolved in a non-aqueous solvent glacial acetic acid and titrated with a strong acid with perchloric acid.

Reagents:

- 0.1N Perchloric Acid in glacial acetic acid (standardized titrant).
- Glacial Acetic Acid (solvent).
- Crystal Violet (indicator).

Experimental Values:

Sample Weight: 150 mg

Actual Normality of $HClO_4(N_{act})$: 0.1005 N

Standard Equivalence Factor (e): 1mL of 0.1N $HClO_4 \equiv 6.006$ mg of Urea

Titration and Endpoint Detection:

Dissolve the 150 mg sample in 40 ml of glacial acetic acid. Add 2 drops of Crystal Violet indicator. Titrate with the perchloric acid until the color changes from violet to blue-green. Perform a blank titration using only the solvent to account for impurities.

Titration Readings:

Sample Titrate Volume (V_s): 24.75mL (3 times Avg)

Blank Titrate Volume (V_b): 0.10mL (3 times Avg)

Net Volume ($V_{net} = V_s - V_b$): 24.56 mL

The purity is calculated using the ratio of the actual normality to the theoretical normality (0.1N)

$$\text{Purity \%} = \frac{(V_s - V_b) \times \text{Equivalence Factor} \times \frac{N_{act}}{0.1} \times 100}{W}$$

$$\text{Purity \%} = \frac{24.65 \times 6.006 \times \frac{0.1005}{0.1} \times 100}{150.0}$$

$$\text{Purity \%} = \frac{24.65 \times 6.006 \times 1.005 \times 100}{150.0}$$

$$\text{Purity \%} = \frac{148.788}{150.0} \times 100 = \mathbf{99.192\%}$$

